

How Nuclear Magnetic Resonance (NMR) Works

Where can NMR be used?

□ Synthetic chemists Drug Discovery and development Life science research □ Food quality and safety **Biochemistry** □ Materials science

RF

What information does it give?

- Most commonly, number of carbons and hydrogens
- □ How the carbons and hydrogens are connected
- □ The electronic environment of the atoms

How else can it be used?

Determine chemical structures □ Identify or quantify components in mixtures

- Quantify molecular compounds
- Monitor chemical reactions



Step 1

The sample is placed in a homogenous field (B_0) that aligns the spins of the nuclei in the sample. The sample sits inside an NMR probe, and the probe is surrounded by the super conducting magnet.





Step 2

The probe contains the radio frequency (RF) coils that transmit strong RF pulses into the sample. These RF pulses generate an additional magnetic field (B_1) . The operator controls the B₁ field through the RF pulses, directing the rotation of the spinning nuclei. The RF coils also detect the much weaker signals produced by the spinning nuclei in the sample.

Step 4

RF signals detected by the probe are sent to the console and then to the host computer, where the user processes them in NMR spectra. The location, shape, and area of the signals in each spectrum provide spatial and connectivity information about the nuclei in the sample.

FID

Free induction decay (FID) is the observable NMR signal generated by non-equilibrium nuclear spin





1H NMR spectra of ethanol (CH₃CH₂OH)

Source: http://materialsscienceandengineering.tumblr.com/post/129245493969/bucoughman-how-nmr-works